

Insights into CO₂ conversion with plasma-electrolysis synergy

Aleksandr Pikalev^{1,2}, Xingyu Chen^{1,3,4}, Vasco Guerra², Guanjun Zhang⁴,
Mauritius C.M. van de Sanden^{(*)1,3}

¹ Dutch Institute for Fundamental Energy Research (DIFFER), Eindhoven 5612AJ, NL.

² Instituto de Plasmas e Fusão Nuclear, Instituto Superior Técnico, Lisboa 1049-001, Portugal.

³ Eindhoven Institute for Renewable Energy Systems (EIRES), and Department of Applied Physics, Eindhoven University of Technology, Eindhoven 5600 MB, NL.

⁴ School of Electrical Engineering, Xi'an Jiaotong University, 710049 Xi'an, China.

(*) m.c.m.vandesanden@diffier.nl

In-situ resource utilization (ISRU) is vital for the human colonization of Mars, in order to be self-sufficient for food and energy production [1]. The Martian atmosphere consists of CO₂ (95.9%), Ar (1.9%) and N₂ (1.9%). This CO₂ abundance can provide a feedstock for direct conversion into O₂ and CO, which can be used for life support and energy conversion respectively.

Amongst the CO₂ conversion and oxygen production technologies, the most promising is the combination of non-thermal plasmas and solid oxide electrolyte cell (SOEC) with an oxygen separation membrane. Although the CO₂ splitting can be performed at the SOEC surface, it requires high temperatures (typically 700-900°C) to break the stable carbon-oxygen double bonds electrochemically. The non-thermal plasma provides a highly-chemical-activated environment favorable for CO₂ dissociation, while the SOEC can separate the O₂ from the plasma environment, limiting the backward reaction of CO combing with O₂ to form CO₂. The plasma-SOEC synergy effects have been reported [2,3], but no detailed investigations of the plasma-SOEC performance enhancement have been conducted.

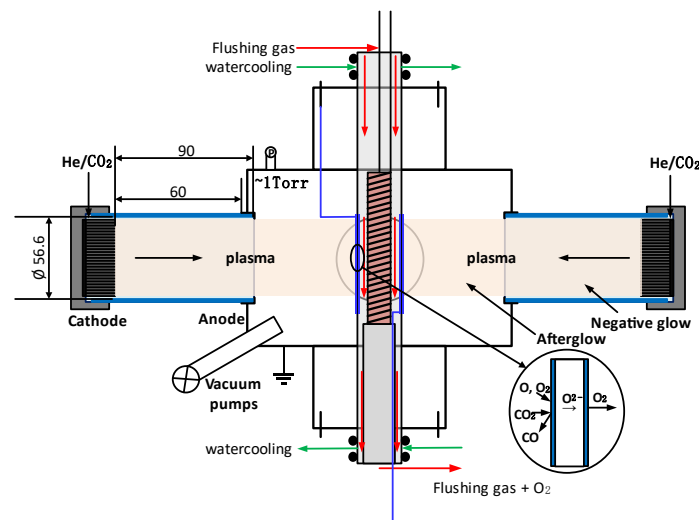


Fig. 1: The scheme of the SOEC integration into the BABE reactor.

To elucidate the kinetics behind the plasma-SOEC synergy, we investigate the plasma-SOEC interactions in the BAri Brush Electrode (BABE) reactor [4] with four DC glow discharges and the afterglow region between their ring anodes. The tubular SOEC consisting of yttria-stabilized zirconia (YSZ) electrolyte and lanthanum strontium manganite (LSM) electrodes is situated in the afterglow region. The scheme of the setup is shown in fig. 1. We investigate several essential factors, including plasma parameters, SOEC working temperatures and bias, and their effects on the catalytic process,

aiming to reach a higher oxygen permeation whilst limiting SOEC working temperature. The CO₂ conversion is studied for He(80%)-CO₂(20%) gas mixture with the pressure of 1 Torr. The plasma conditions are investigated in the discharge region by means of optical emission spectroscopy and in the afterglow region using a Langmuir probe. The dissociation products are determined by mass-spectrometry.

The oxygen pumping experiments demonstrate that the glow discharge can enhance the oxygen transport at the low SOEC working temperatures (350–650°C). At 298W discharge power and 10 sccm feed gas flow, the oxygen permeation can be enhanced from ~0.07mln/min to ~0.6mln/min. Simultaneously, the oxygen separation enhances the CO₂ conversion from 87.1% to 89.7%. The OES shows that the oxygen pumping out of the afterglow region also reduces the intensities of the atomic oxygen lines in the discharge region. In our conditions, the plasma-induced enhancement of the oxygen permeation grows with increasing plasma power, reaching saturation at ~300W. The results are compared with the experiments with He-O₂ and pure He experiments [5].

This work was partially supported by the European Space Agency under Project I-2021-03399. IPFN activities were funded by FCT (Fundação para a Ciência e a Tecnologia) under projects UIDB/50010/2020, UIDP/50010/2020, LA/P/0061/202 and PTDC/FIS-PLA/1616/2021 (<https://doi.org/10.54499/UIDB/50010/2020>) (<https://doi.org/10.54499/UIDP/50010/2020>) (<https://doi.org/10.54499/LA/P/0061/2020>) (<https://doi.org/10.54499/PTDC/FIS-PLA/1616/2021>). The authors are thankful to Ashley Hughes for critical reading of the abstract.

- [1] V. Guerra et al. *J. Appl. Phys.* **132** (2022) 070902.
- [2] A. Pandiyan et al. *J. CO₂ Utilization* **57** (2022) 101904.
- [3] F. Buck et al. *J. Industrial and Engineering Chem.* **104** (2021) 1–7.
- [4] S. Ratynskaia, G. Dilecce, and P. Toliás *J. Plasma Phys.* **81** (2015) 345810202.
- [5] X. Chen et al. *ICPIG XXXV*, Egmond aan Zee, The Netherlands, 2023, 132.