Quantitative analysis of NO² generated in Atmospheric Pressure Plasma Jet using Ion Mobility Spectrometry

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Reactive oxygen and nitrogen species (RONS) play key roles in numerous biochemical, physiological, and pathological processes. Among them, species such as O_3 , OH, NO₂, NO₃, N₂O₅, etc., can be efficiently generated utilizing atmospheric pressure plasma jets (APPJ) in Argon [1–4]. The controlled production of RONS by APPJ finds widespread applications in biomedicine for wound healing [5], cancer treatment [6], immune cell activation [7], surface treatment and sterilization [8, 9]. Various diagnostic techniques, including Fourier-transform infrared (FTIR) spectroscopy [10], chemical and semiconductive gas sensors [11, 12], have been employed for RONS detection.

The APPJ utilized in this study was developed at the Department of Experimental Physics, Comenius University Bratislava [1]. The Ar plasma jet was generated within a glass capillary with an internal diameter of 0.5 mm, employing a hollow needle to cylinder geometry configuration. A high voltage with an amplitude from 1 to 4 kV (0.09 to 1.2W) and a frequency of 9.45 kHz was applied to the needle. The flow rate of Ar (purity 4.6) was varied at 20, 50, 100, 150 and 200 mL/min.

The IMS employed for detection of NO₂ generated by APPJ and assessing electron transfer reactions between reactant ions (RI) O_2 . CO_2 . $(H_2O)_{0,1,2}$ and NO_2 was developed by MaSa Tech Company. The IMS is a rapid (milliseconds to seconds range) [13], highly sensitive (parts per billion to parts per quadrillion level) [14], good portable and powerful analytical technique useful in various fields [15]. Notably, NO₂ and other RONS can be detected in negative polarity mode.

The IMS spectra of the APPJ at discharge power ranging from 0.09 to 1.2 W and Ar gas flow of 100 mL/min are shown in the Fig. 1. The dominant peak with reduced mobility of 2.23 cm².V⁻¹ s⁻¹ represents RI. Peaks with reduced mobility of 2.50 and 2.38 cm².V⁻¹ s⁻¹ represent ions NO₂⁻ and N₂O₂⁻, respectively. The ionization of neutral NO_x species was carried out using negative chemical ionization method, with RI. In the earlier study, was the IMS quantitative calibrated for NO₂ species [16]. The production of NO₂ generated in APPJ increased with increasing of discharge power (Fig. 1). The maximum NO₂ production of 5.3 ppm was observed at power of 1.2 W (4 kV) and Ar gas flow rate of 100 mL/min.

Fig. 1: The IMS spectra of the APPJ at discharge power ranting from 0.09 to 1.2 W and Ar gas flow of 100 mL/min.

The dependence of the concentration of $NO₂$ generated in the APPJ at different discharge powers, an Ar flow rate of 100 ml/min and different distances of the IMS from the APPJ is shown in the Fig. 2. The concentration of $NO₂$ slowly increased with increasing IMS-APPJ distance. The maximum $NO₂$ production (5.3 ppm)was observed at a distance 5 mm and decreased with further increasing IMS-APPJ distances.

Fig. 2: Production of NO₂ generated in the APPJ at different discharge powers, Ar flow 100 mL/min and different distances of the IMS from the APPJ $(0 - 50 \text{ mm})$.

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References:

- [1] G. Horváth *et al., IEEE Trans. Plasma Sci.* **41(3)** (2013) 613–619
- [2] I. Jõgi *et al.*, *Contrib. Plasma Phys.* **60(3)** (2020) e201900127
- [3] R. Brandenburg *et al.*, *IEEE Trans. Plasma Sci.* **37(6)** (2009) 877–883
- [4] B. C. Adhikari *et al., Results Phys* **30** (2021) 104863
- [5] S. Bekeschus *et al., Clin Plasma Med* **4(1)** (2016) 19–28
- [6] Y. Li *et al., Sci Rep* **7(1)** (2017) 45781
- [7] A. Lin *et al.*, *Int J Mol Sci* **18(5)** (2017) 966
- [8] P. Lamichhane *et al.*, *React Chem Eng* **5(11)** (2020) 2053–2057
- [9] O. V. Penkov et al., *J Coat Technol Res* **12(2)** (2015) 225–235
- [10] H. Yi *et al.*, *Atmos Meas Tech*, **14(8)** (2021) 5701–5715
- [11] S. Fischer *et al.*, *Sens Actuators B Chem* **147(2)** (2010) 780–785
- [12] M. T. Vijjapu *et al.*, *ACS Sens* **5(4)** (2020) 984–993
- [13] M. Sabo *et al., Talanta* **85(1)** (2011) 400–405
- [14] V. A. Kostarev *et al.*, *Talanta* **245** (2022) 123414
- [15] G. A. Eiceman and Z. Karpas, *Ion Mobility Spectrometry* 2nd edition. CRC Press (2005) 350
- [16] E. Maťaš *et al., Eur. Phys. J. D* **77(2)** (2023) 21